





First trials in electrokinetic remediation of heavy metals from a contaminated marine dredged sediment from a European port

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4th April 2019, Dubrovnik







Characterization of the sediment



CaCO₃(g/kg)	Total organic carbon (TOC) (g/kg)	Solids concentration (mass %)	Agglomerated mass (g)	Dry matter (mass %)	Microtox solid phase test
10,2	21	33	0	34,85	>16 g/L of sediment

Organic compounds	mg/kg
Total Organostannic compounds	<0,09
Total Polycyclic Aromatic Hydrocarbons (PAH)	<0,48
Total Petroleum Hydrocarbons (TPH): C10-C40	70
Total Polychlorinated Biphenyl (PCB): 28, 52, 101, 118, 138 153, 180	<0,007

Equilibrium tests



- Desionized water
- Nitric acid 1M
- Sodium hydroxide 1M
- Citric acid 0,2M
- Acetic acid 0,5M
- Oxalic acid 0,2M
- EDTA 0,2M
- Potassium ioide 0,2M





Electrokinetic remediation trials





Test	Anolyte	Catholyte	Direct Electrical field (V)
1	Water	Water	No
2	Water*	Water	12, 20, 30
3	Water	Acetic acid	12

 Samples: pH, conductivity, voltage, intensity, electroosmotic flux, metals and Infrared spectroscopy

- *During the last week, samples taken were replaced by H_2O_2
- Tests 1 and 2: 3 weeks, Test 3:1 week

Maximum % extraction of metals in test 2



Conclusions

- Metal extraction in equilibrium tests: concentrated HNO₃ for Cu, Zn and Pb. NaOH for As. EDTA for Cr.
- Metal extraction in electrokinetic remediation trials: acetic acid at 12V > extraction than water at 30V.
- Design of the remediation process should incorporate the particularities of the different contaminants in their interaction with the extracting solutions and the characteristics of the sediment.

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