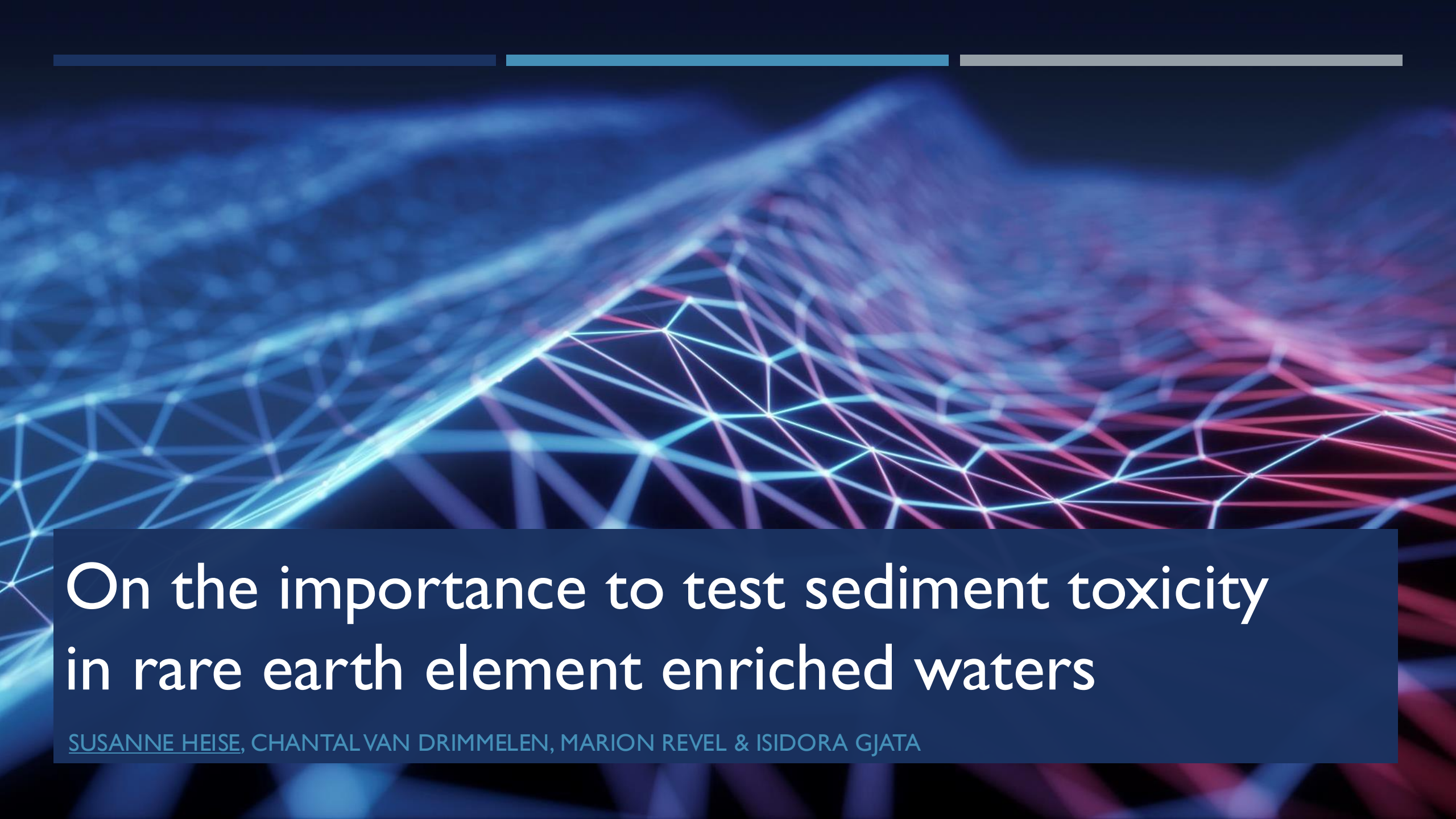




On the importance to test sediment toxicity in rare earth element enriched waters

SUSANNE HEISE, CHANTAL VAN DRIMMELEN, MARION REVEL & ISIDORA GJATA

Structure

- What are rare earth elements (REE)?
- Why do they occur in waters at elevated concentrations?
- What determines their fate in the environment?
- Are they toxic?
- Why should sediments be in focus?
- What are the next steps?

Rare Earth Elements (REE)

Rare Earth Elements by Geology.com																	
H																	He
Li	Be											B	C	N	O	F	Ne
Na	Mg											Al	Si	P	S	Cl	Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
Cs	Ba	La-Lu	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
Fr	Ra	Ac-Lr	Rf	Db	Sg	Bh	Hs	Mt									
Lanthanides																	
		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	
Actinides																	
		Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr	

A group of 17 chemically similar elements:
Lanthanides plus yttrium and scandium

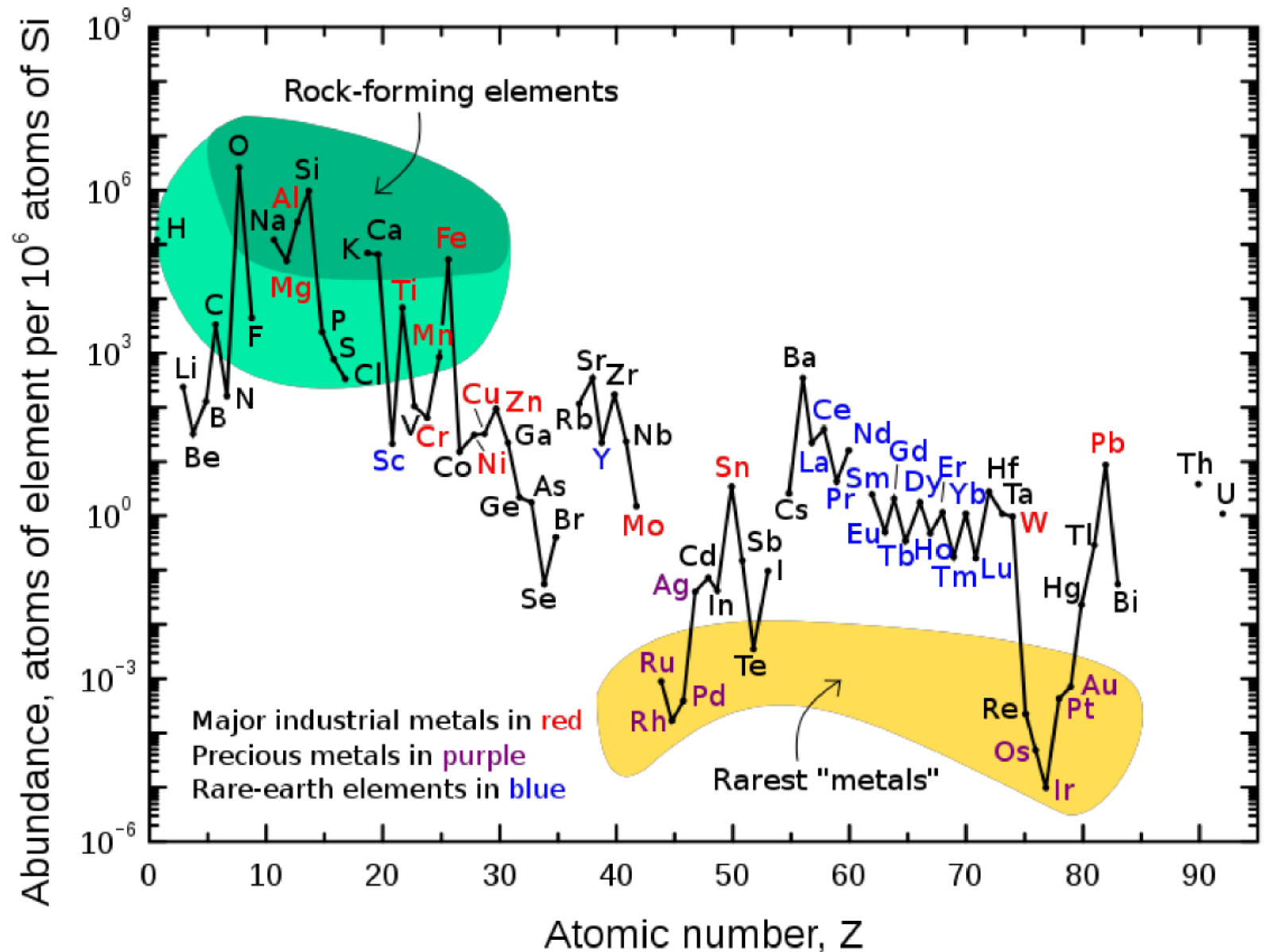
Are Rare Earth Metals rare?

Not really.

→ widespread

→ but occur in low concentrations.

→ Technology Critical Elements (TCE)



Reported anthropogenic anomalies in water bodies

River / Water Bodies	Country	REE	Source
Weser	Germany	Gd	Kulaksiz & Bau, 2007
Rhine	Germany	La, Sm, Gd, Ce	Kulaksiz & Bau, 2013
Atibaia River	Brazil	Gd	de Campos & Enzweiler, 2016
San Francisco Bay	USA	Gd	Hatje et al. 2016
Hérault River	France	Gd	Rabiet et al. 2009
Teltow Channel	Germany	Gd	Knappe et al. 2005
Wupper	Germany	Gd	Bau & Dulski 1996
Rhine Estuary	The Netherlands	Nd, Pr, Sm	Bakkenist & van de Wiel 1995
Tagus Estuary	Portugal	Y	Brito et al. 2018
Tokio Bay	Japan	Gd	Nozaki et al. 2000

Applications of Rare Earth Elements

Unique properties:

Magnetic (Nd, Pr, Dy)

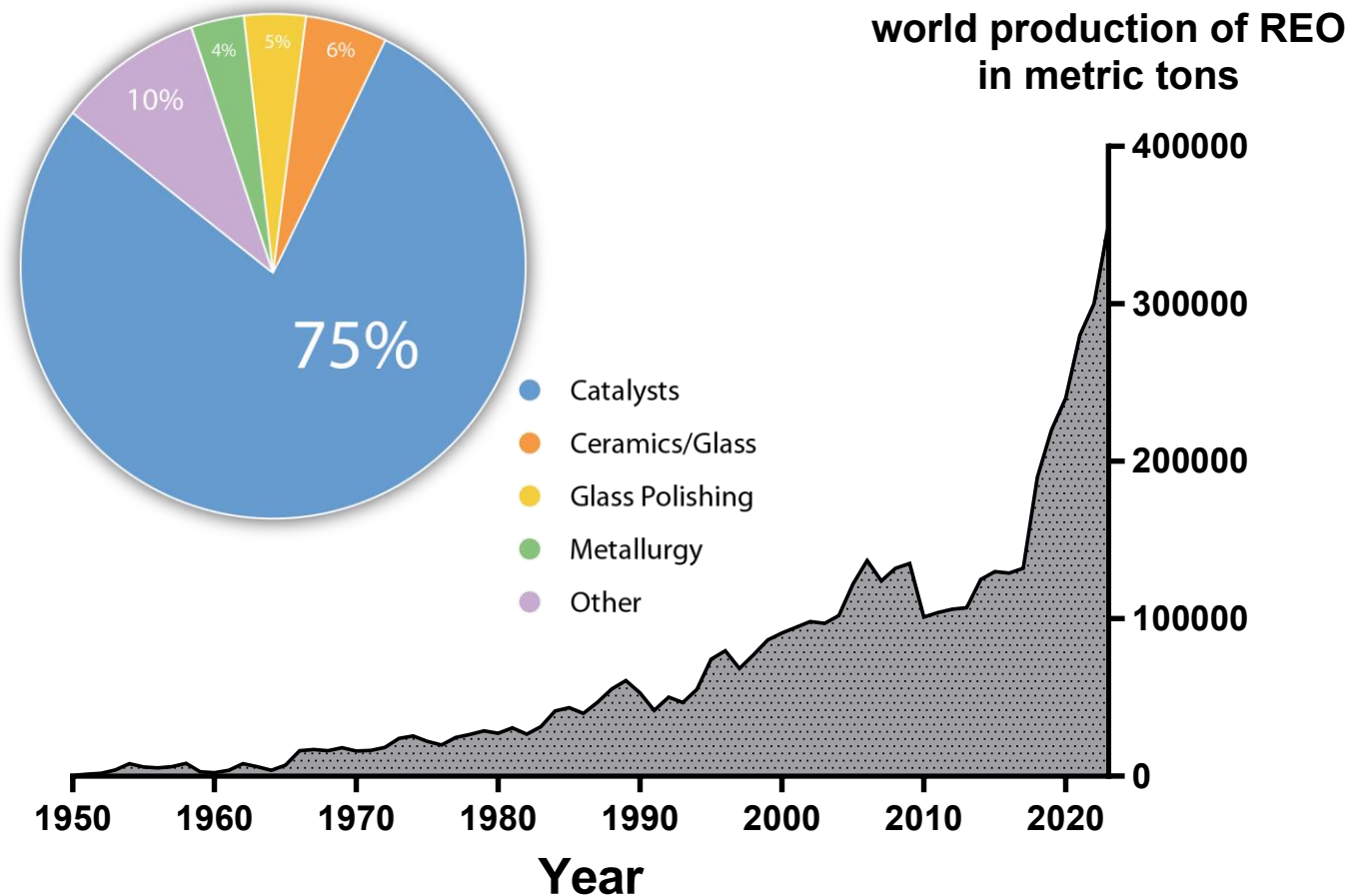
Optical (fluorescing – e.g. Eu)

Catalytical (La, Ce)

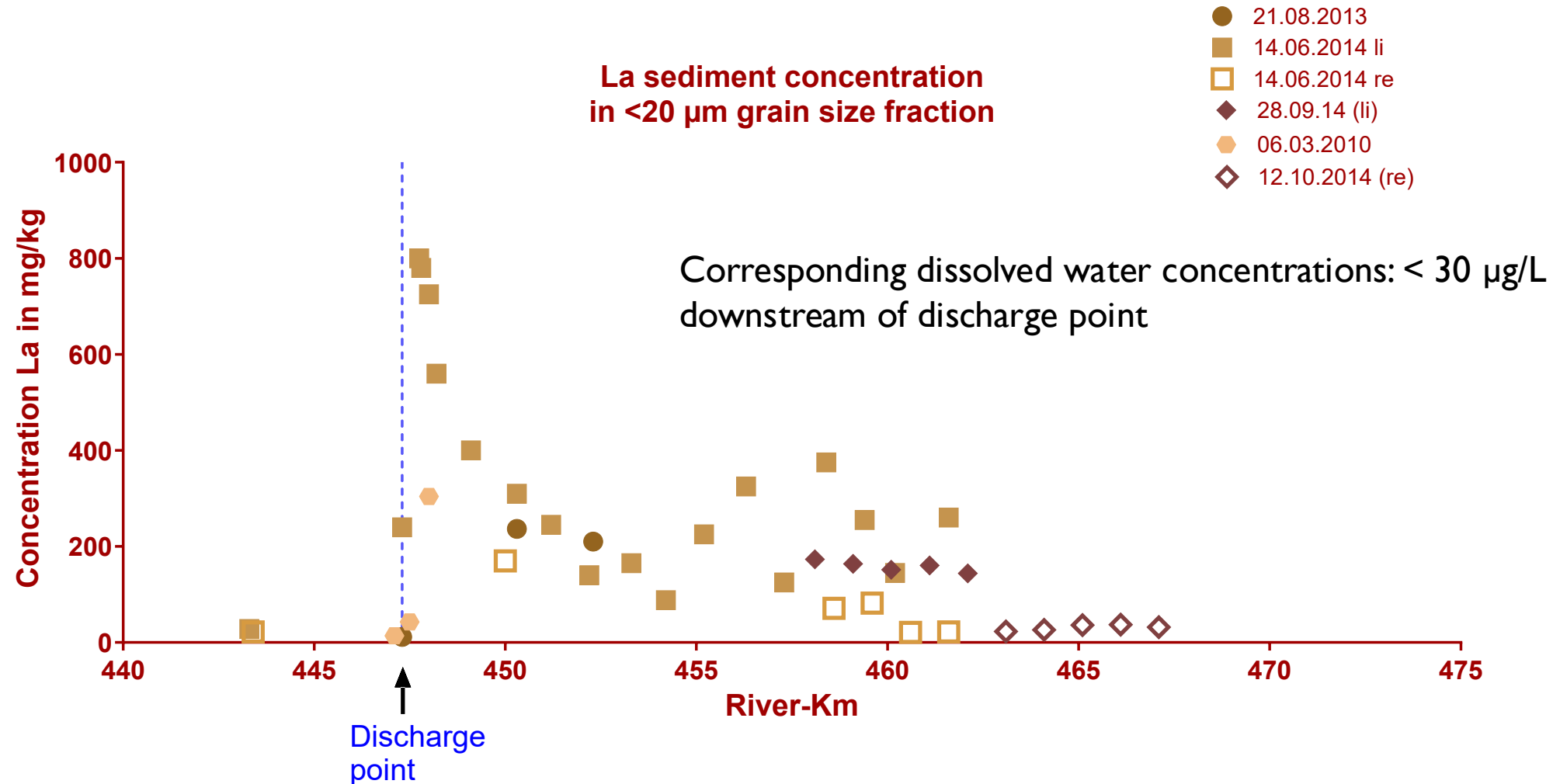
Thermic (Sm)

.....

Increasing concern regarding the effects of rising environmental concentrations.
(Revel et al. 2024)



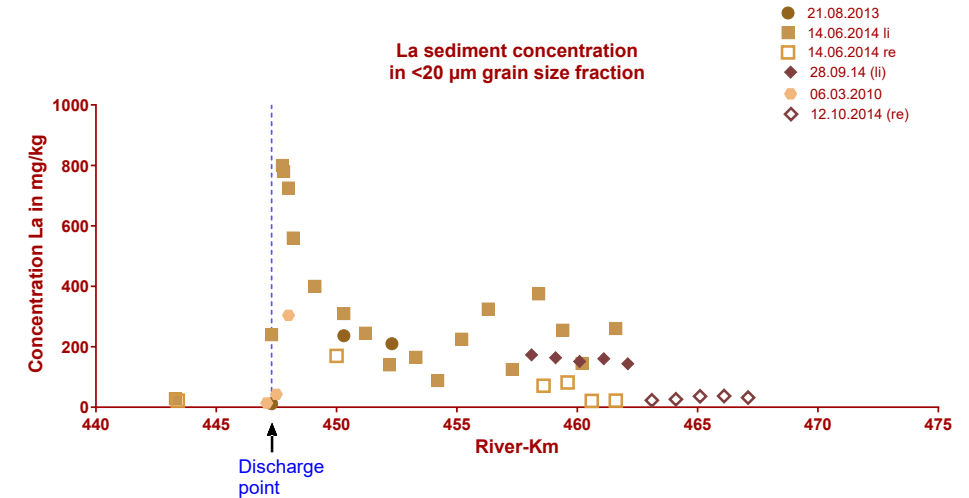
Industrial emissions of La into the Rhine River



Environmental fate of REE in the aquatic systems

[Ln³⁺] sinks with

- ↑ pH (formation of Ln-hydroxides)
- ↑ organic matter
- ↑ alkalinity (formation of Ln-carbonates)
- ↑ PO₄-concentrations (formation of Ln-phosphates)



Affinity to sediments is high;

K_p-Values (sediment-pore water distribution) is much higher than for other metals.

Most REEs in the aquatic system are found in suspended form or sediment-bound.

Interaction of REE with biological systems

Lanthanum blocks Ca-channels (among others) in the cells

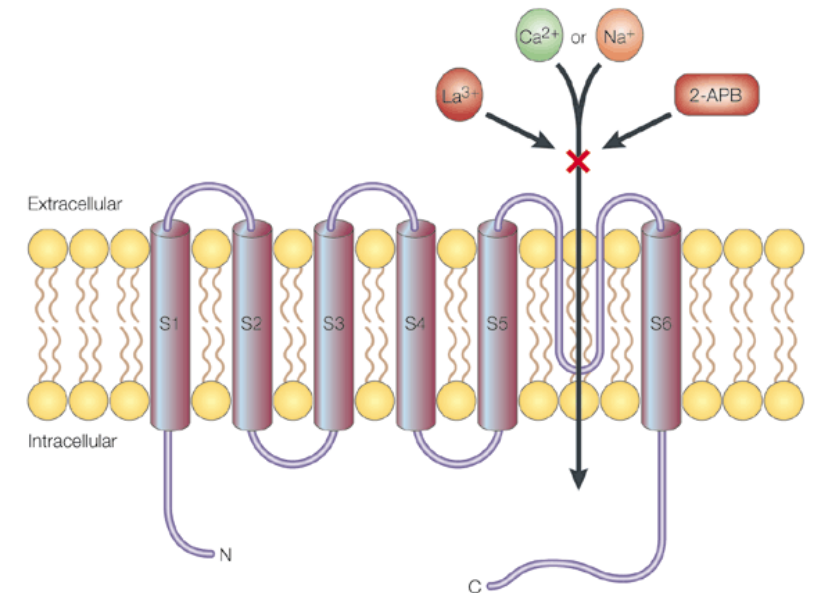
Ca^{2+} : Essential element in all cells hydrated ionic radius 0.26 nm (Kiriukhin & Collins 2002)

Ln^{3+} (octahydrates): 0.23 – 0.25 nm (Rudolph & Irmer 2020)



Calcium-Antagonist with a variety of potential physiological consequences
(signalling, cell protection, development)

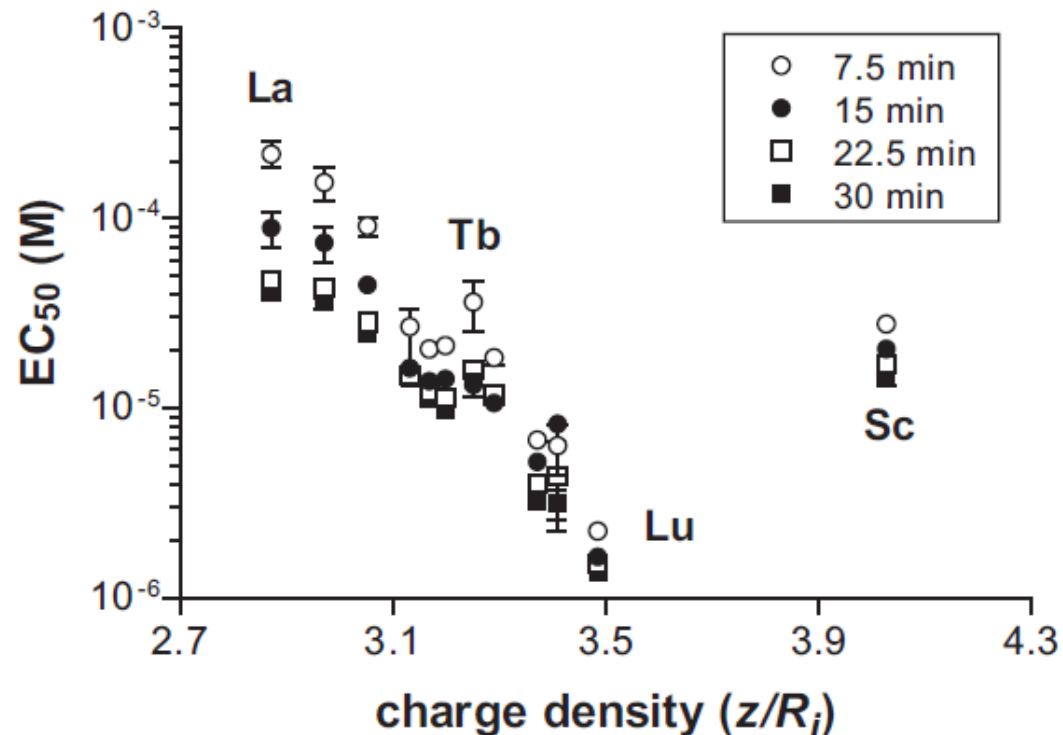
BUT: How toxic is „toxic“?



Are they toxic?

Bacteria

EC₅₀ values of 11 REE and Sc
when exposed to *Aliivibrio fischeri* („Microtox“)



Results from Weltje, 2002:

- Toxicity of Ln comparable to that of Cd, Cu, Pb, Zn
- Toxicity seems to increase with atomic number

Weltje, L. (2002). "Bioavailability of Lanthanides to Freshwater Organisms. Speciation, Accumulation and Toxicity. PhD-Thesis, Delft University."

Acute daphnia toxicity

Control



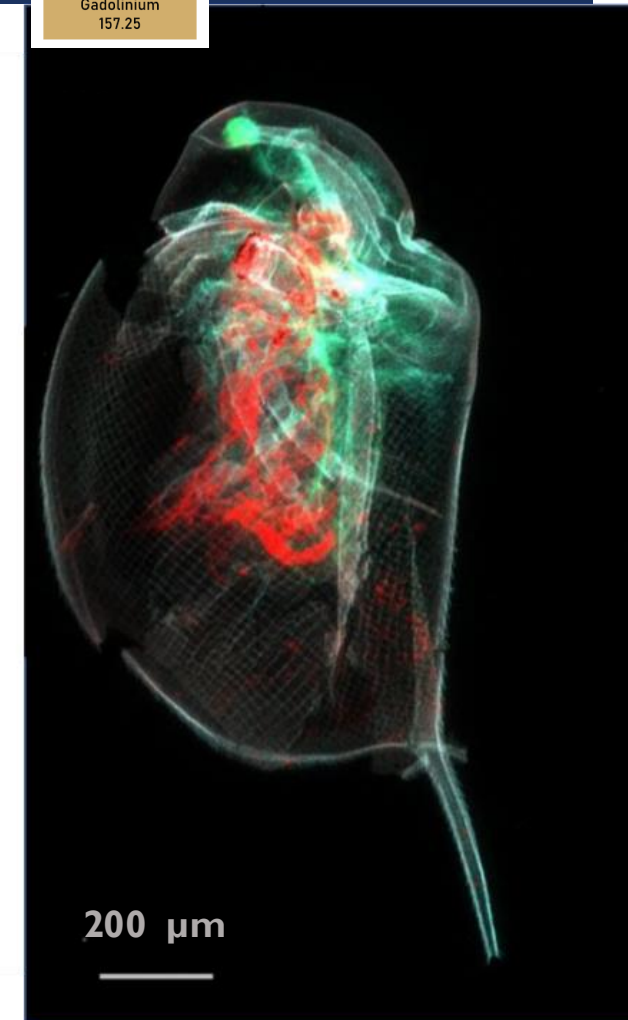
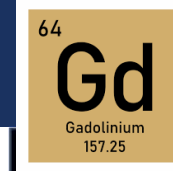
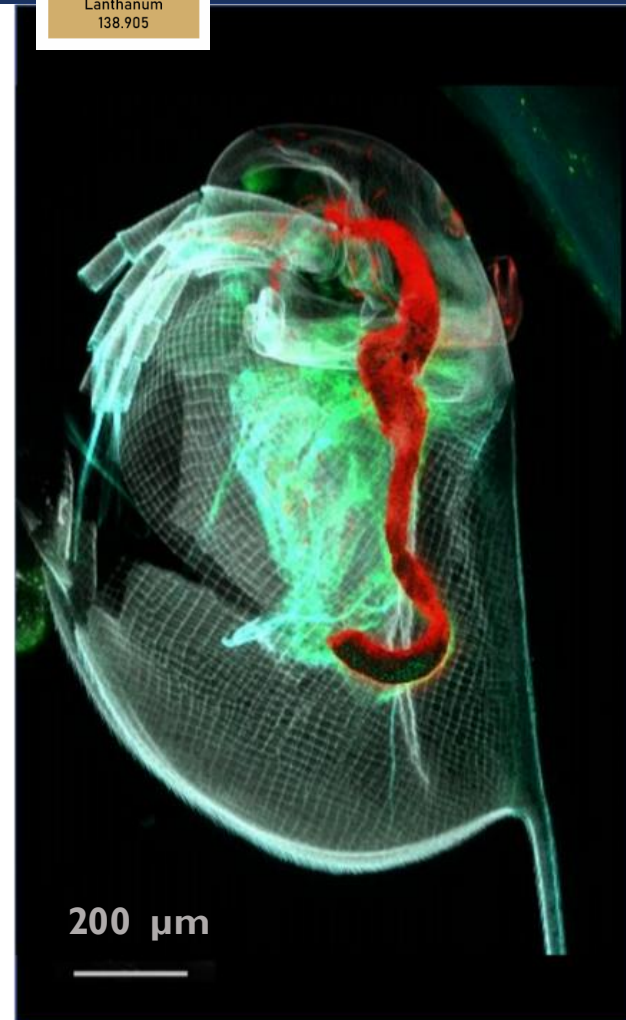
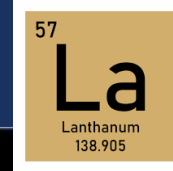
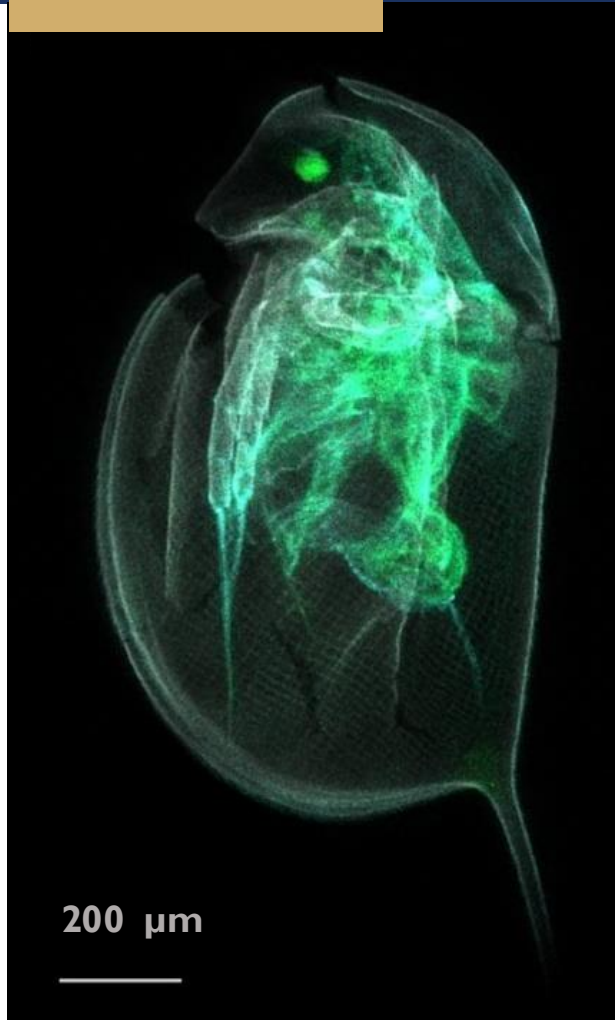
Grey: calcium

Green: sulfur

Cyan: phosphorus

Red: lanthanide

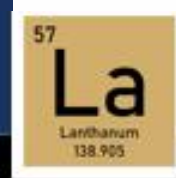
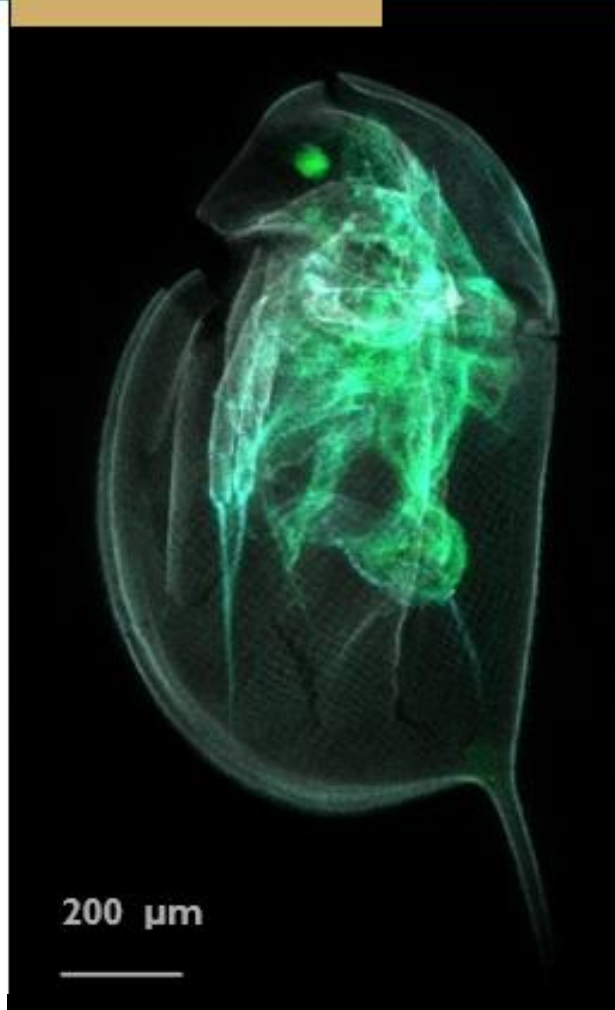
(Revel 2023)



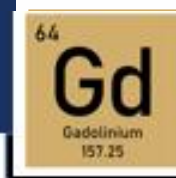
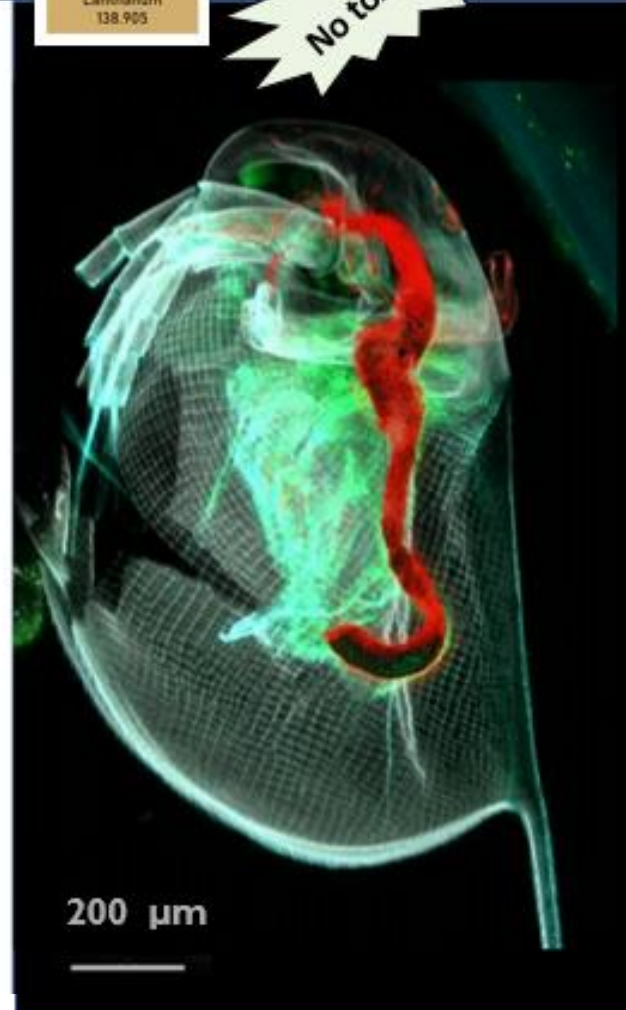
Acute daphnia toxicity



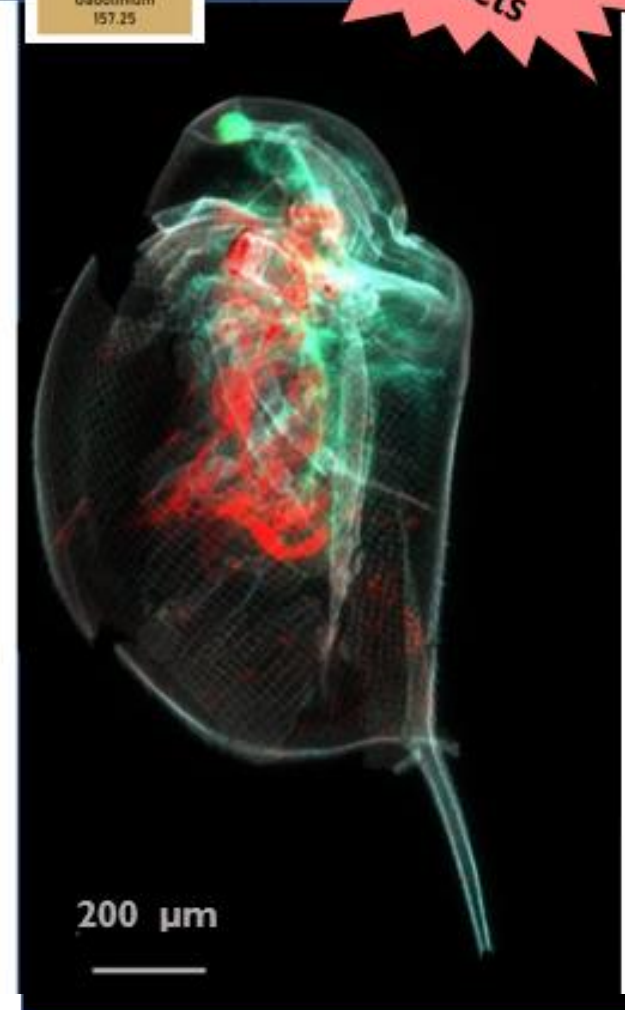
Control



No toxicity



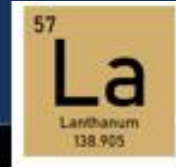
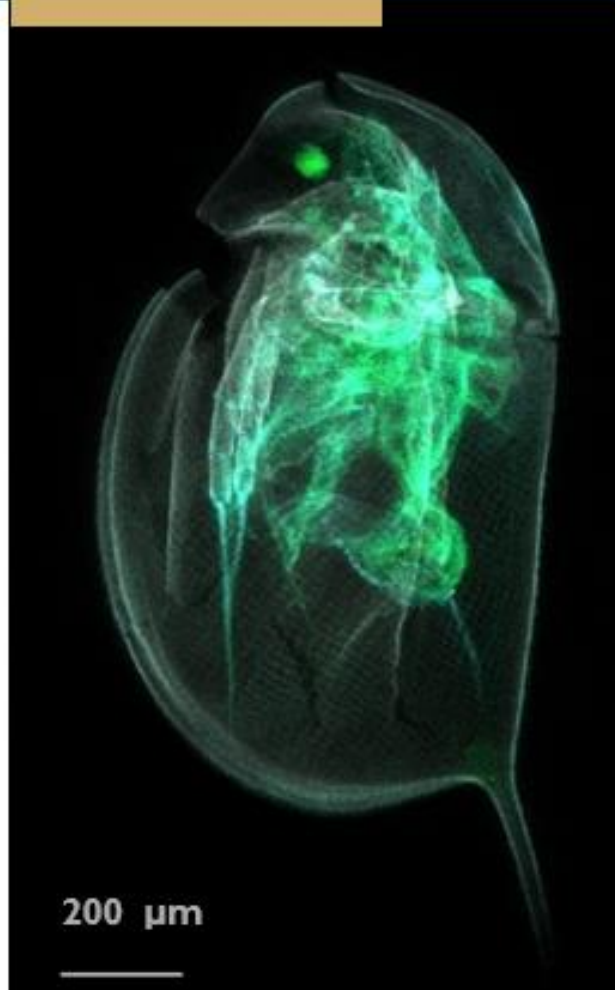
Lethal effects



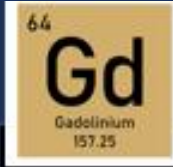
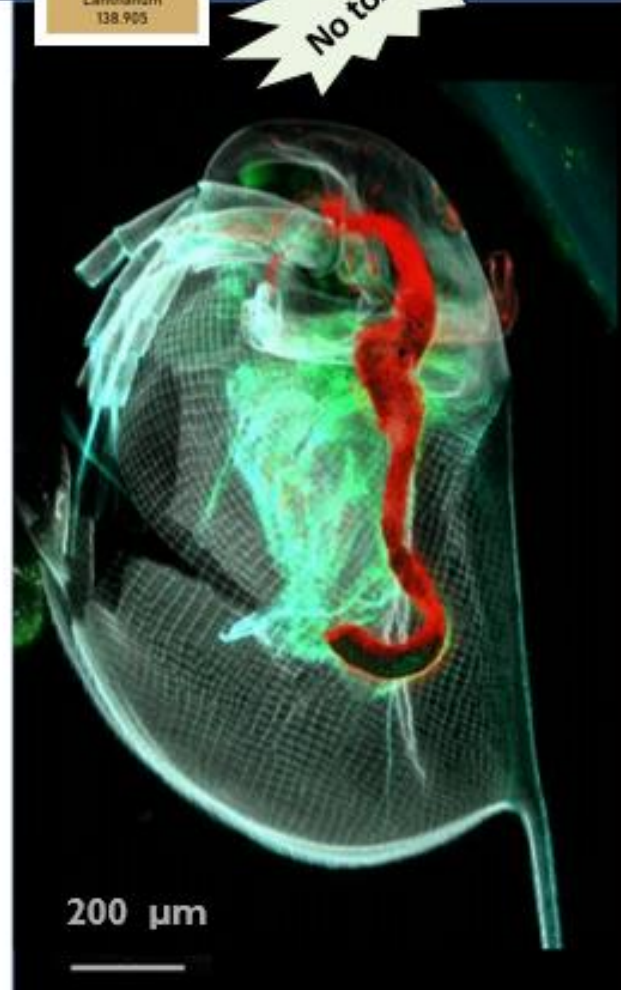
Acute daphnia toxicity



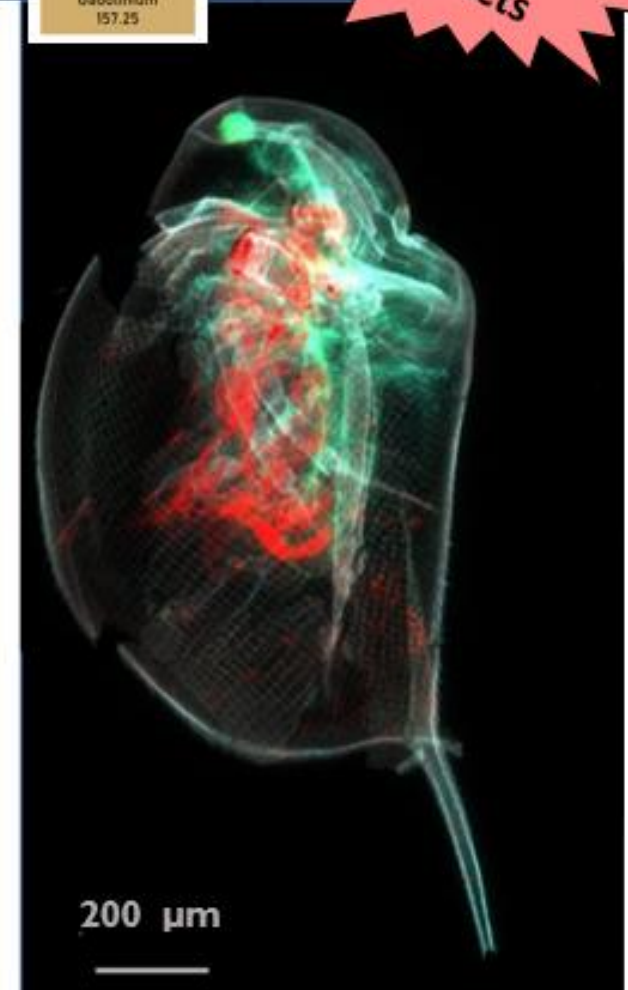
Control



No toxicity



Lethal effects



Acute daphnia toxicity



Gd: higher dissolved concentrations than La

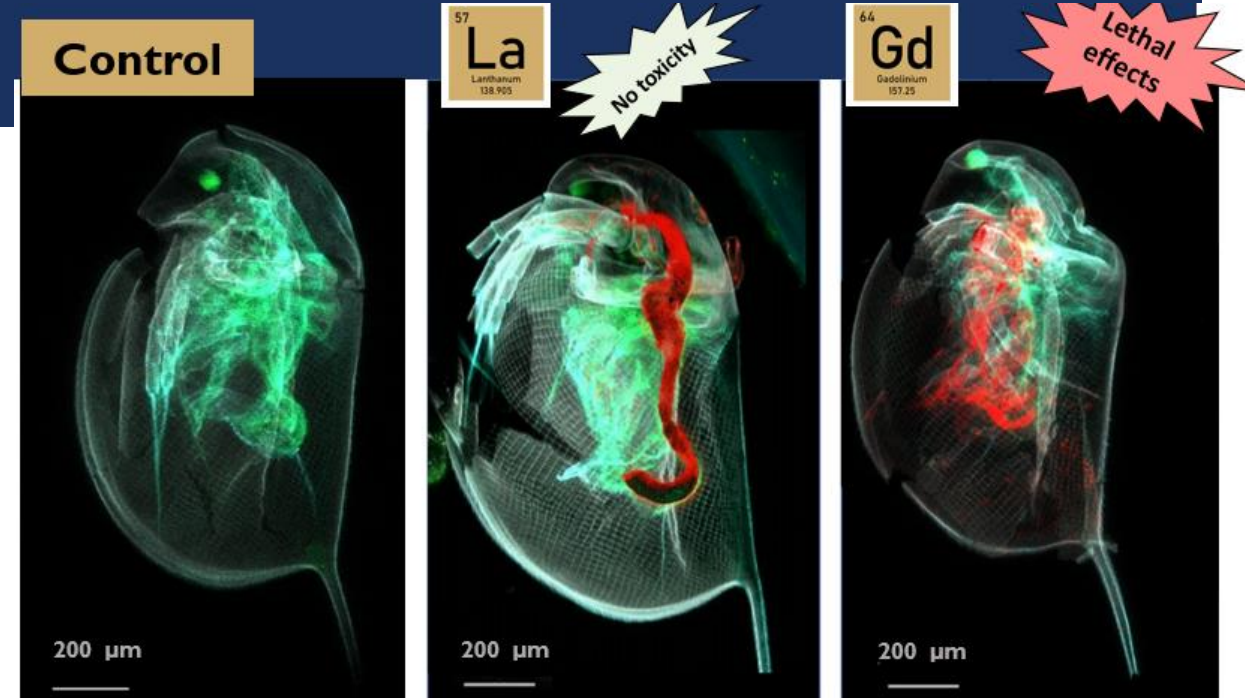
→ distributed throughout the tissue, more toxic

↔ La: probably precipitates, accumulated (and restricted) to GI

Effects on daphnids under acute exposure

IC₅₀ (Gd): 13.84 mg/L (11.21 – 23.84 mg/L)

IC₅₀ (La): > 30 mg/L

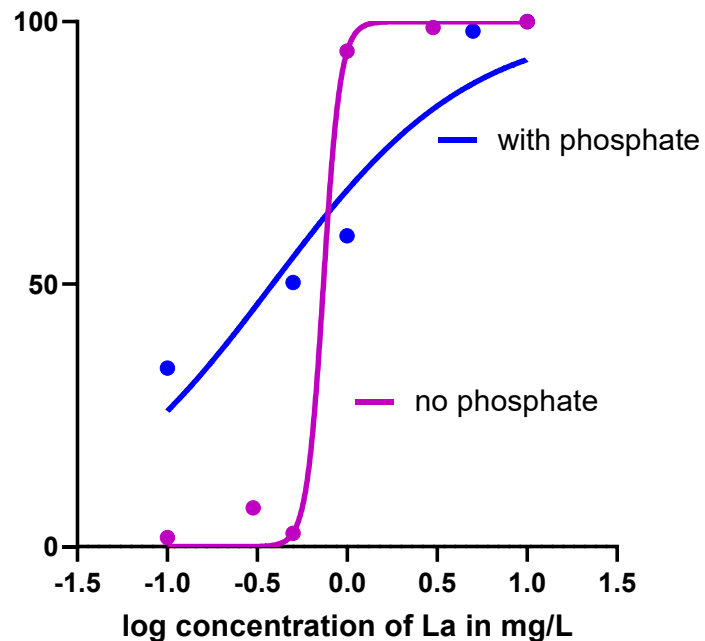


>> than current water concentrations
Toxicity caused by free ions?
Precipitates are not toxic?

Toxicity despite precipitates...

Comparison of dose-response curves for La in nematode bioassay (EC_{50} : 0.38 – 0.6 mg/L) in media with and without phosphate

Inhibition of reproduction
in % of control



In the presence of phosphate, no La should be available as free ion, but there is a clear dose-response curve, similar EC_{50} .



[9]

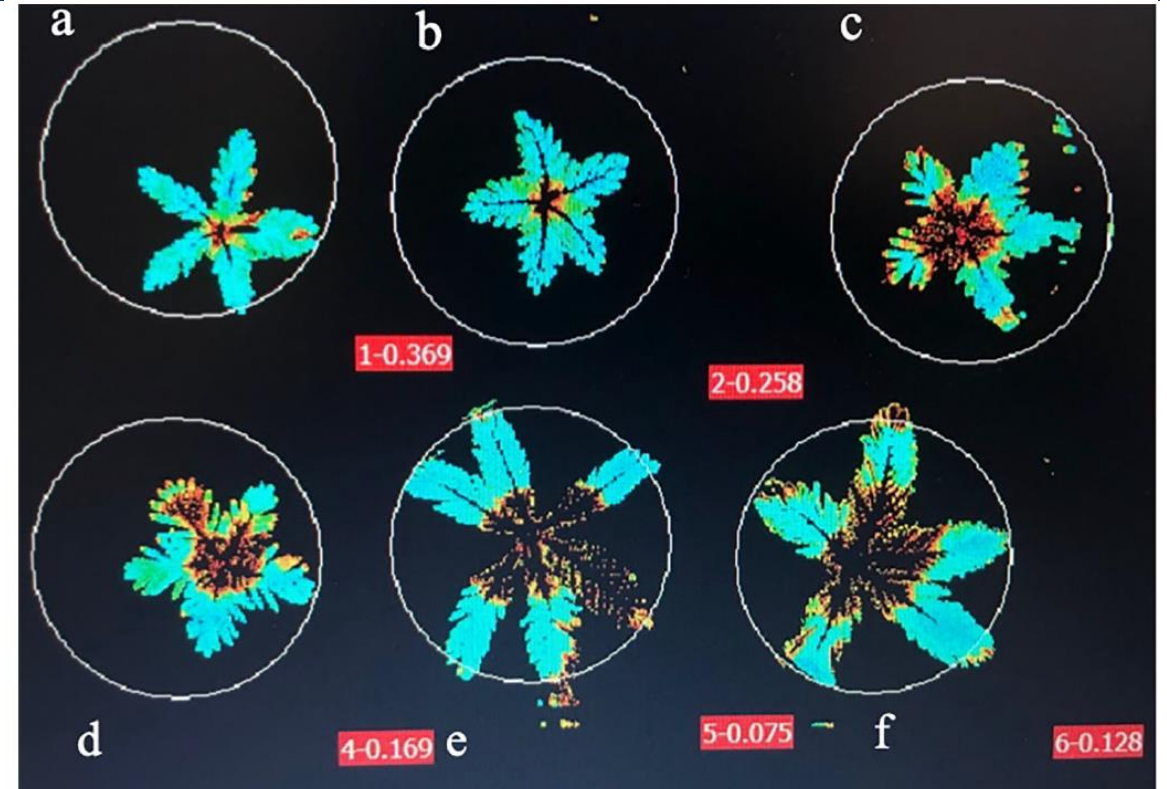
Uptake of solid phases and remobilization in the gut system?

Sediment toxicity of Gd to a benthic plant



Negative impact of Gd over 10 d

- Growth significantly reduced at 250 mg/kg
- Photosynthetic yield reduced at 500 mg/kg



Chlorosis/Necrosis in *Myriophyllum aquaticum* at different Gd-concentrations in sediments

Summary

- REE have a high affinity to sediments
- With increased use, concentrations in sediments will rise.
- Indications, that precipitates and sediment concentrations will harm aquatic organisms
- The current number of ecotoxicity tests for sediment-bound REE is too small to conclude

More data on sediment toxicity are needed, before REE should be assumed to be non-hazardous to aquatic life.

Exposure matrix

